



*Unité
Sciences du Bois et des Biopolymères*



Chemical Functionalization of Lignocellulosic Polymers for the Control of Interfacial Adhesion in Wood and BiofiberPlastic-composites

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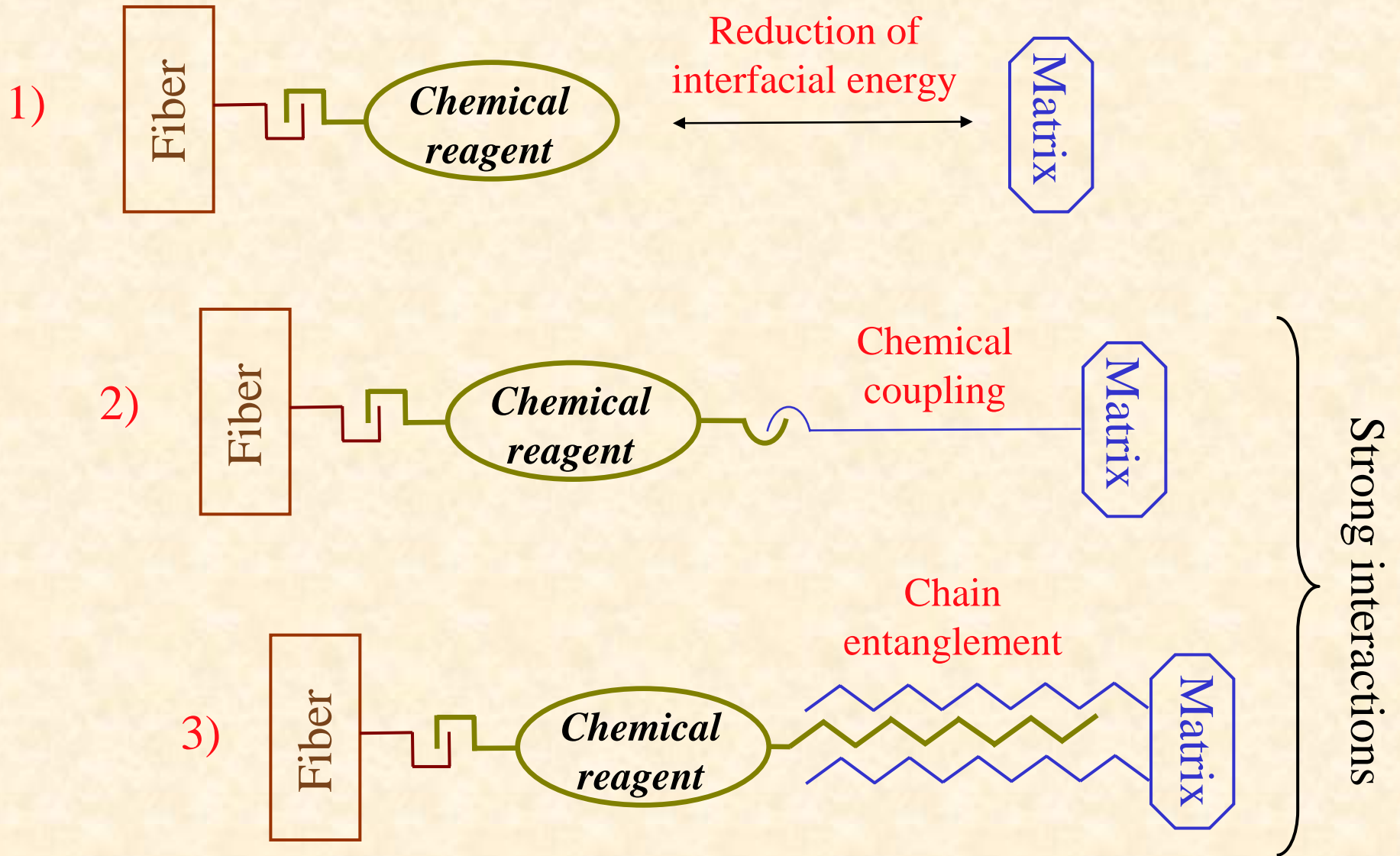
INTRODUCTION

- ❑ The mechanical properties of Wood or BiofiberPlastic composites (WPC) are strongly influenced by the quality of the fiber/matrix interface
- ❑ The poor adhesion between the hydrophilic lignocellulosic fibers and the hydrophobic polymeric matrix limits the strength performances of the WPC's

However,

- ❑ The interfacial adhesion in WPC's can be improved via the chemical **functionalization** of the lignocellulosic **fibers**

PRINCIPLE



ADDITIONAL IMPROVEMENTS

□ The properties of the **lignocellulosic fibers** can be additionally improved if the functionalization is performed in the bulk:

- Dimensional stability
- Decay resistance
- Photostability
- Water repellency
- Thermoplasticity (if the grafted chain is long enough)

OBJECTIVES OF MY RESEARCH

□ To **develop** novel chemical pathways for the functionalization of lignocellulosic materials and to **characterize** the modifications at a molecular level:

- Reactions with **alkoxysilanes**
- **Transesterification** reactions

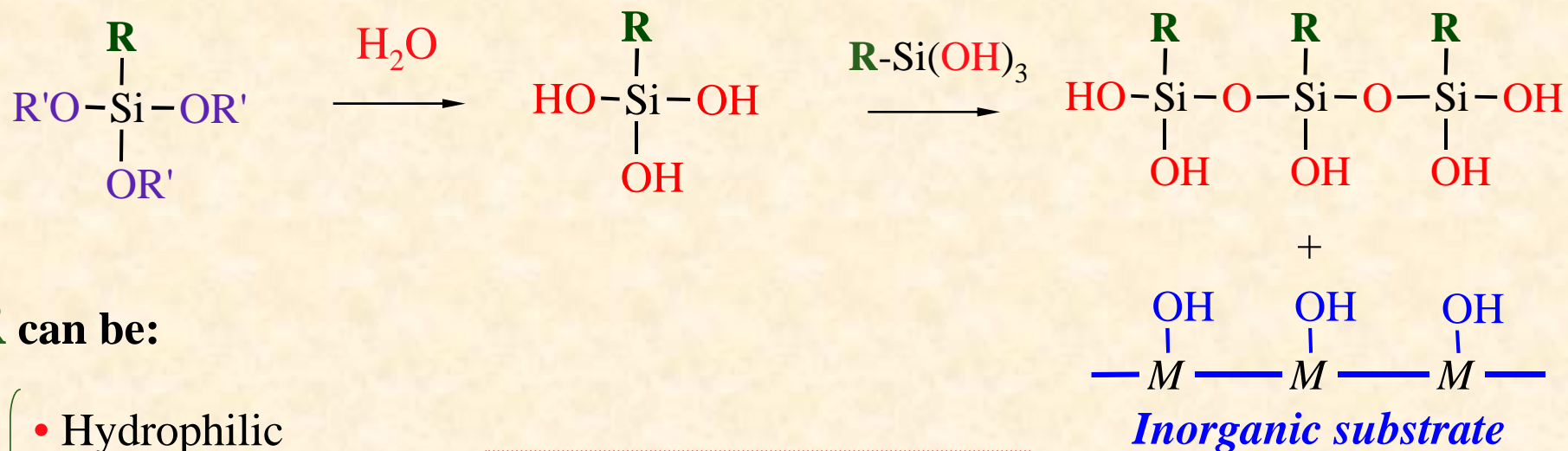
*My presentation
today*

□ To study the **impact** of these modifications on the properties of **composites** made of wood or plant fibers

□ To **modelize** the mechanisms involved in the interfacial region

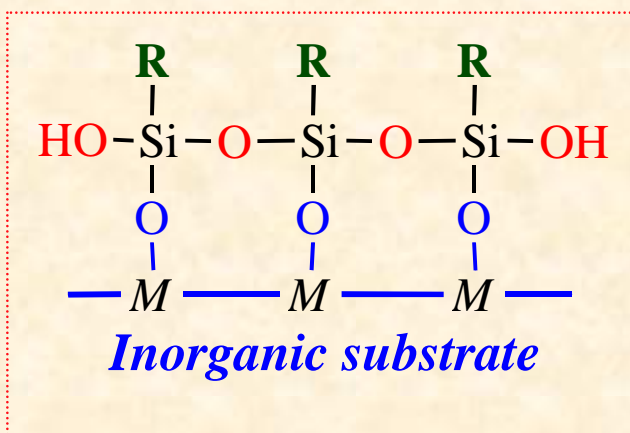
REACTION WITH TRIALKOXYSILANES

□ Trialkoxysilanes are reactive towards most inorganic substrates:

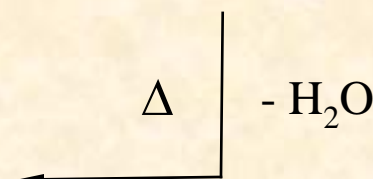


R can be:

- Hydrophilic
- Hydrophobic
- Organoreactive
- UV or heat resistant...

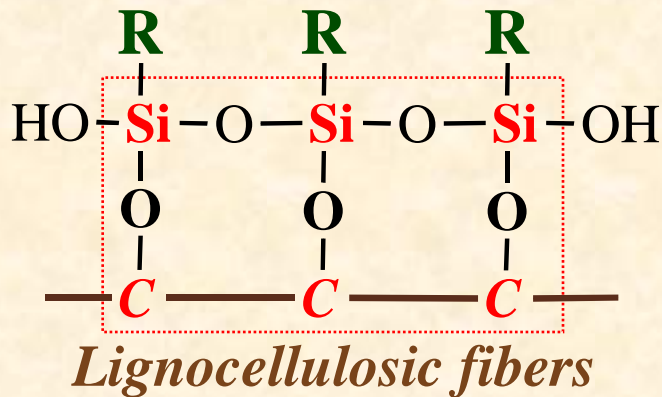


(M = Si, Al,...)



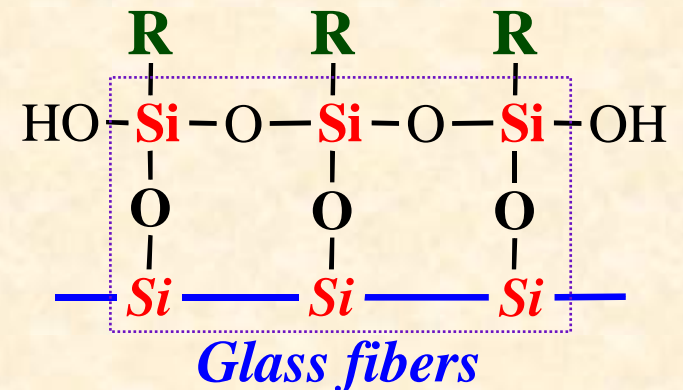
REACTION WITH TRIALKOXYSILANES

- Similar reactions can be envisaged with lignocellulosic substrates, **but** in that case, the silane bond is **not stable**



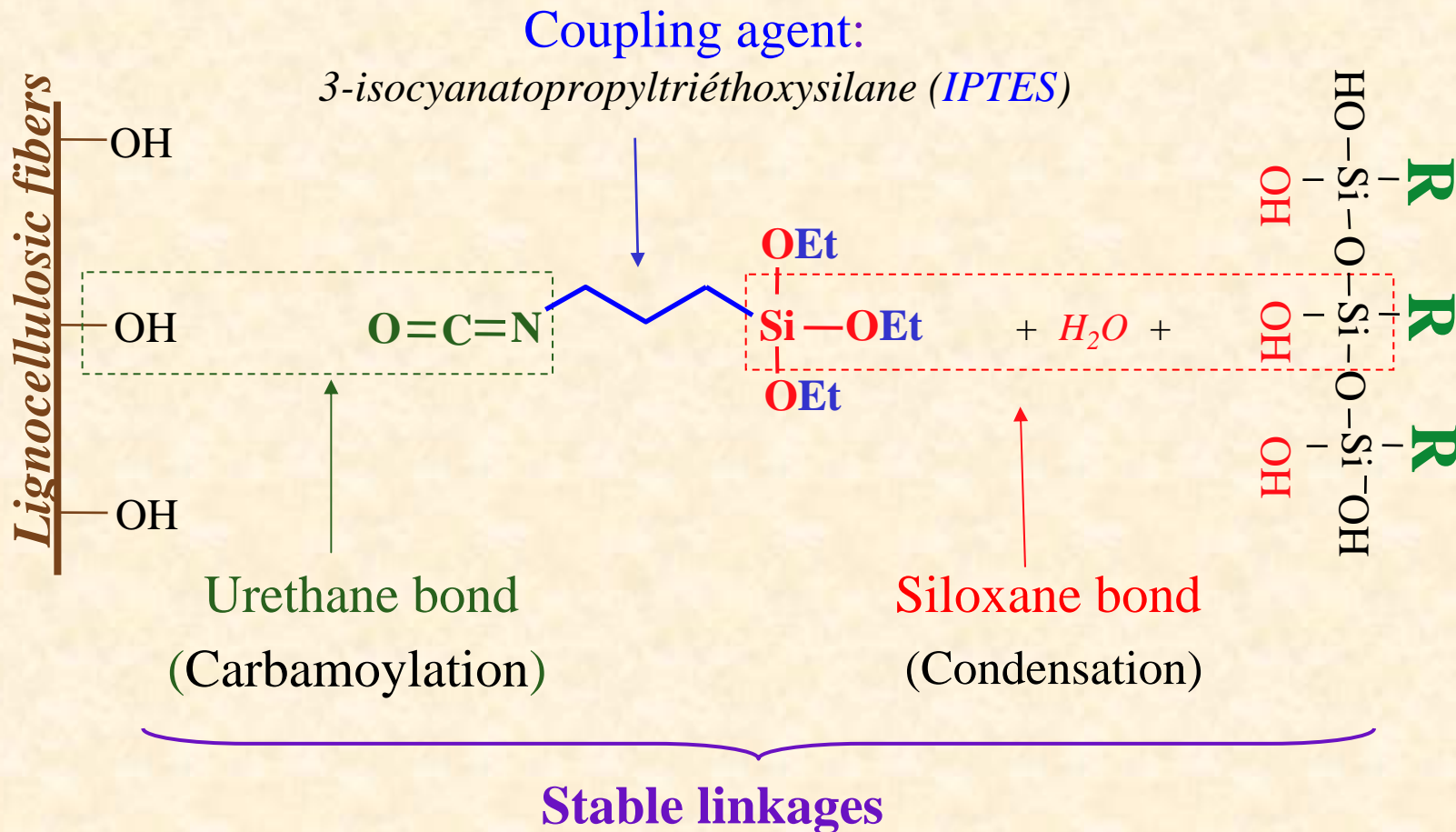
← **Hydrolysis** in the presence of water

But **stable** siloxane bonds with
siliceous substrates



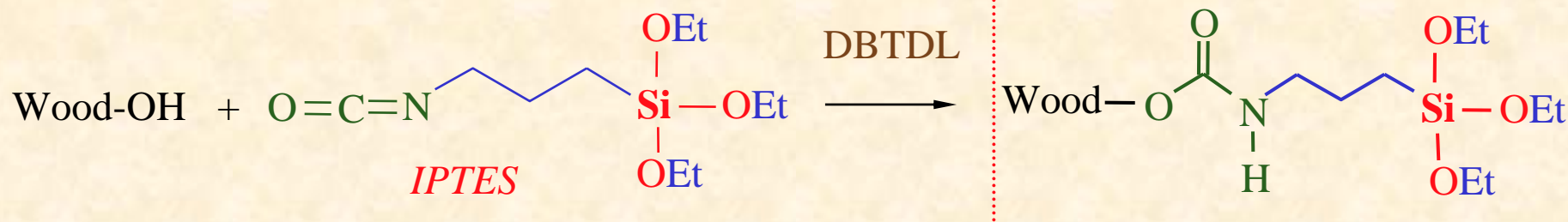
OUR APPROACH

- Utilization of an organofunctional silane as a coupling agent

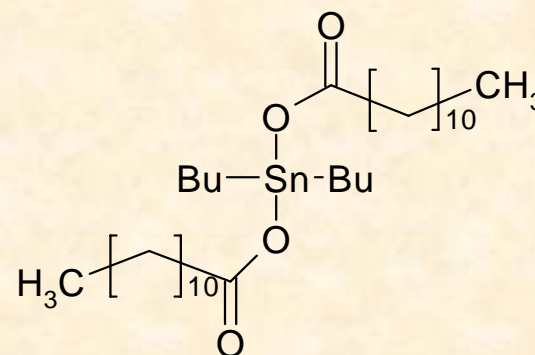


REACTION BETWEEN *IPTES* AND WOOD

□ Reaction



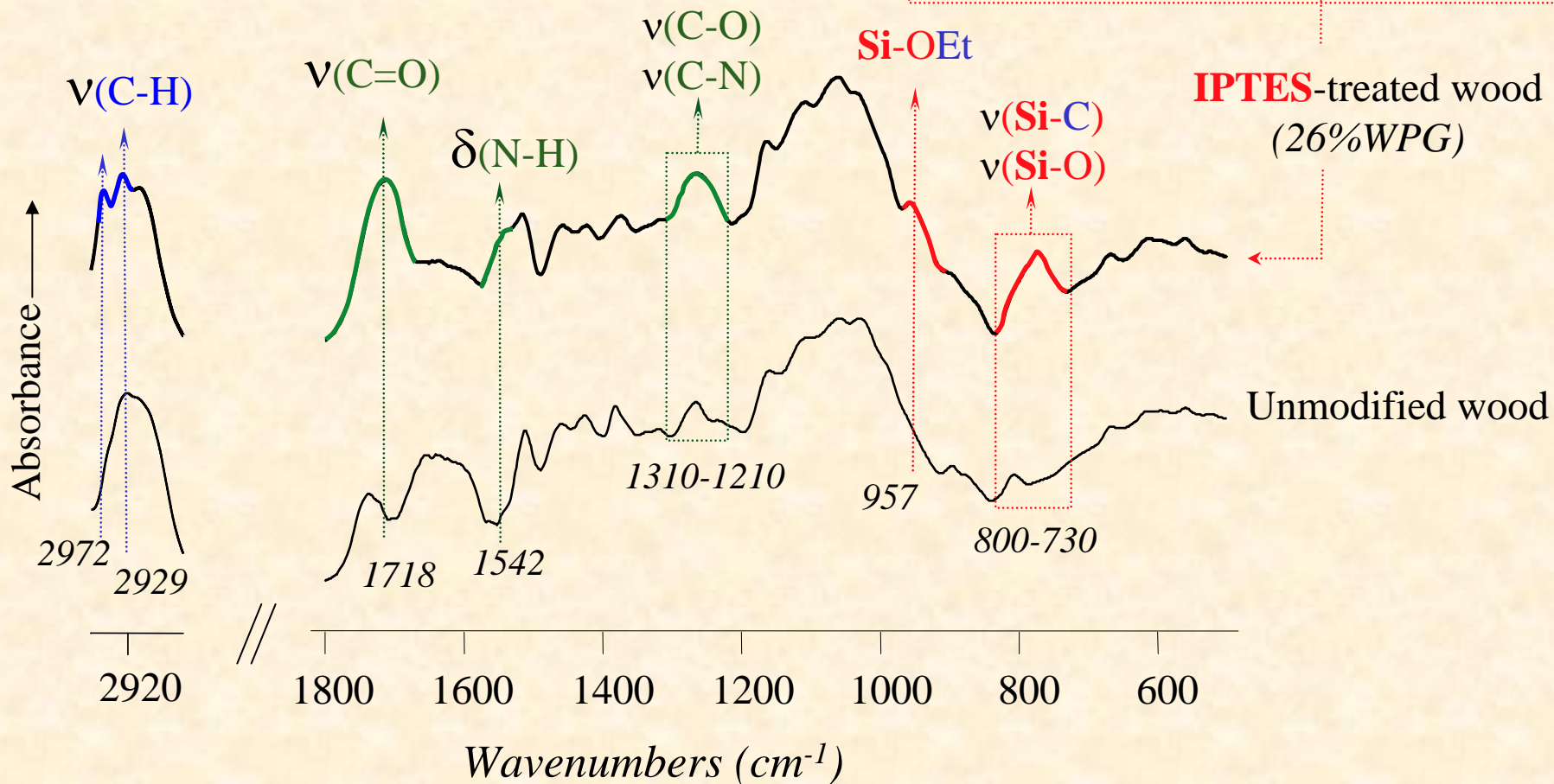
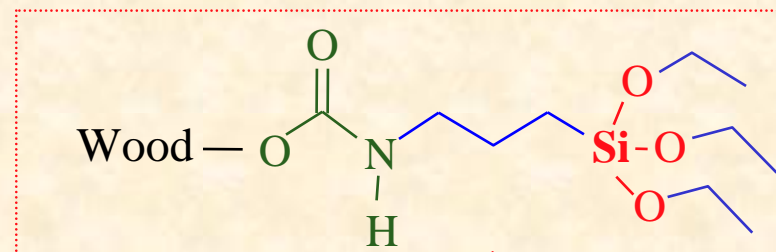
Catalyst = Dibutyltin dilaurate (DBTDL):



- Different modification rates were investigated and characterized by FTIR and solid-state **NMR** spectroscopy.

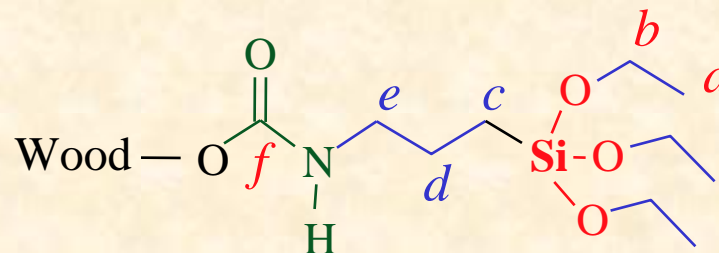
REACTION BETWEEN *IPTES* AND WOOD

□ FTIR spectroscopy

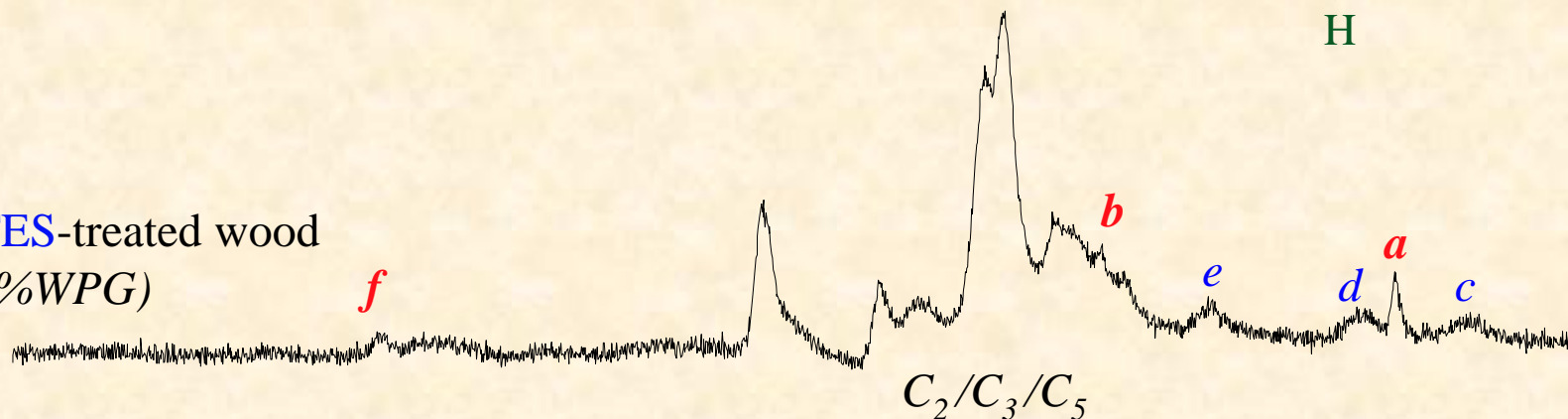


REACTION BETWEEN IPTES AND WOOD

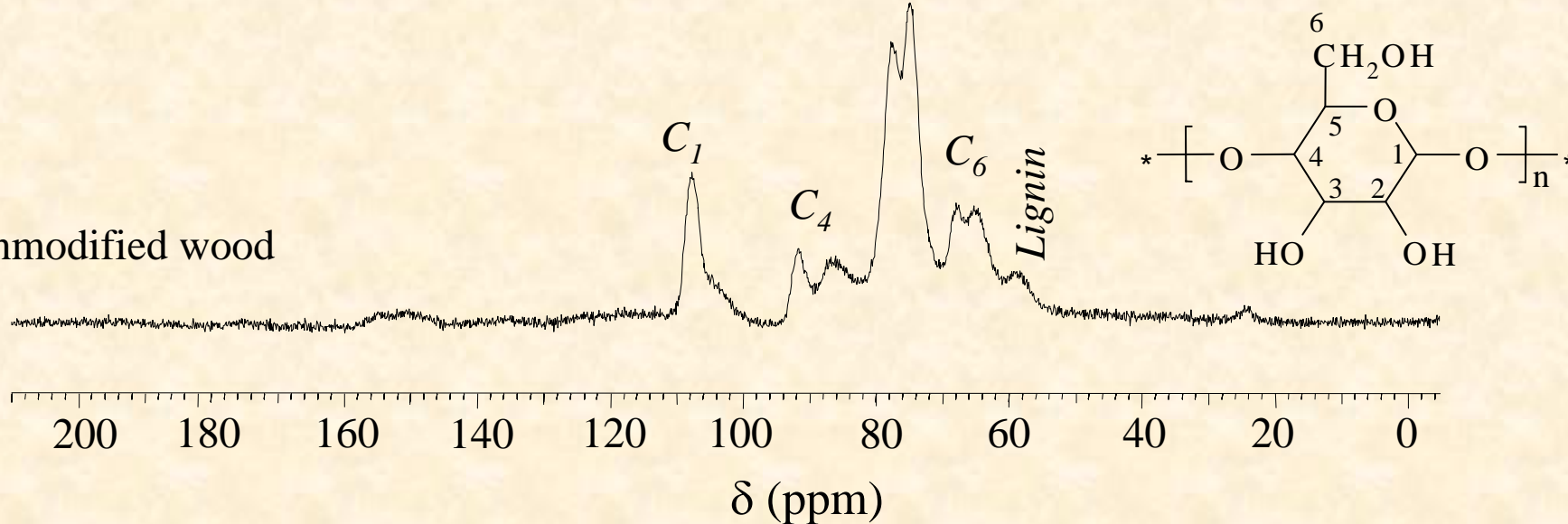
□ ^{13}C CP-MAS NMR spectroscopy



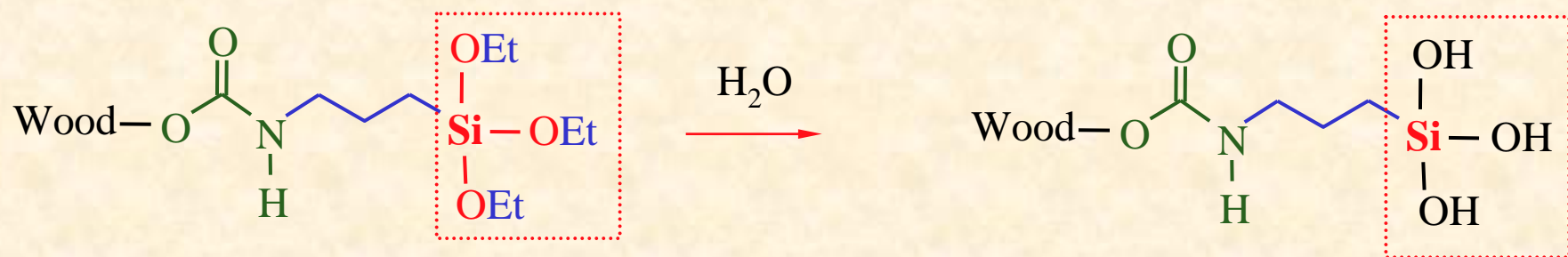
IPTES-treated wood
(26%WPG)



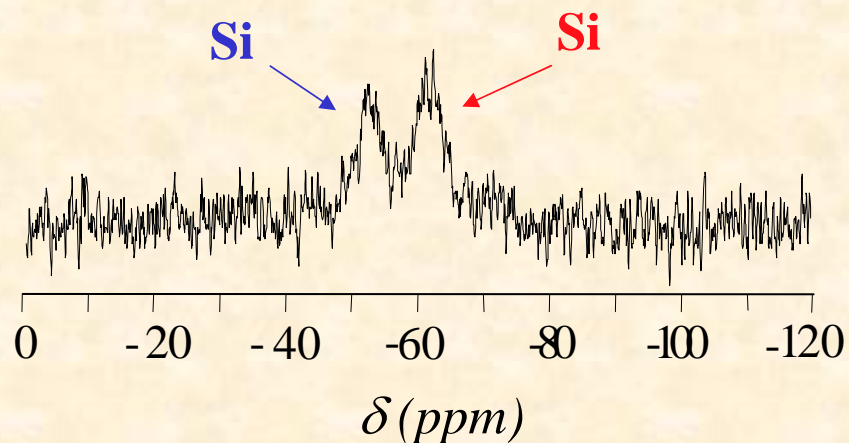
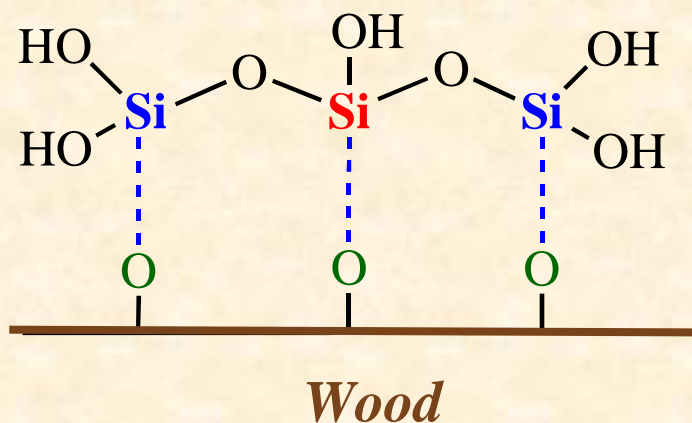
Unmodified wood



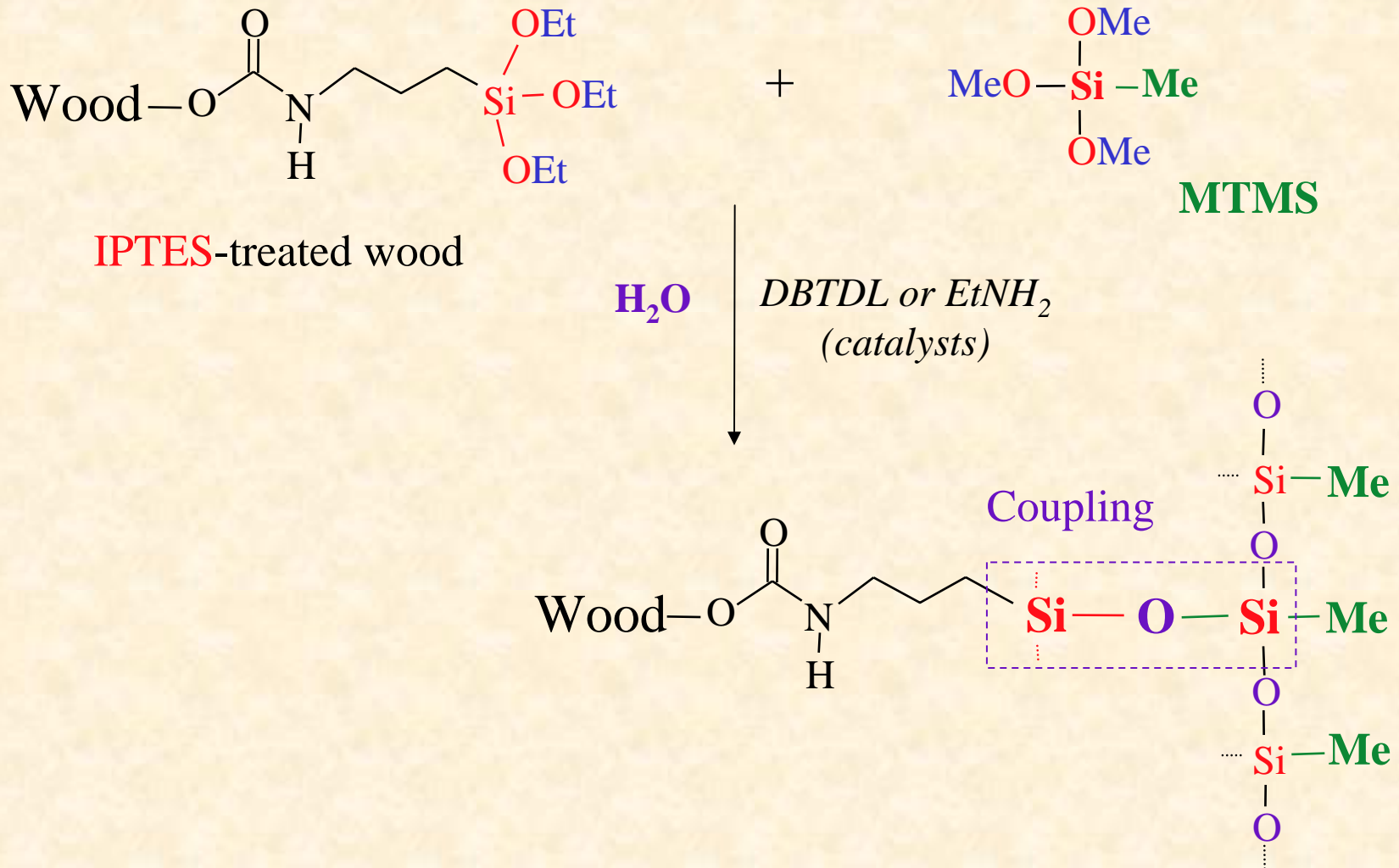
HYDROLYSIS OF *IPTES*-TREATED WOOD



- Cleavage of the **Si-OEt** linkages confirmed by FTIR and ¹³C NMR
- Partial condensations between neighboring **Si-OH** (²⁹Si NMR)

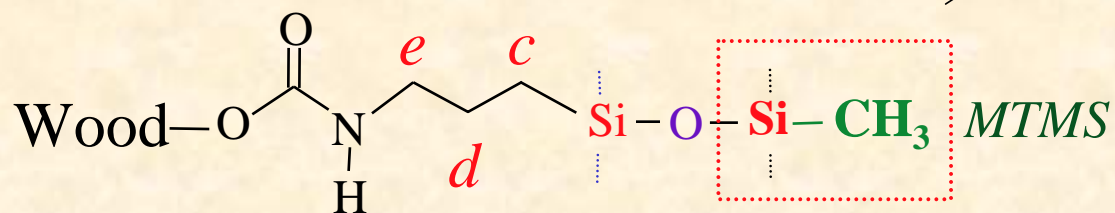


COUPLING WITH METHYLTRIMETHOXYSILOXANE (MTMS)

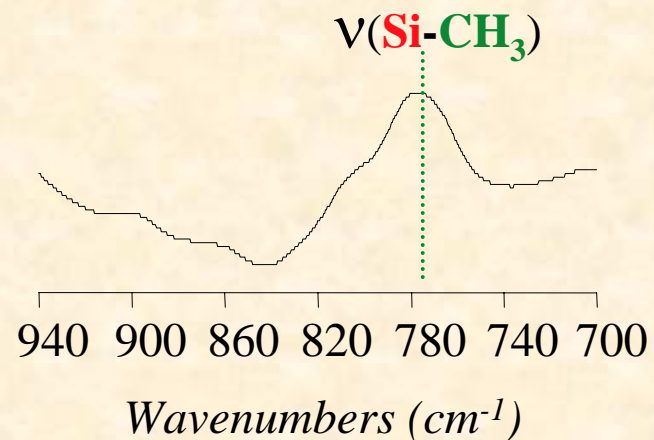


COUPLING WITH MTMS

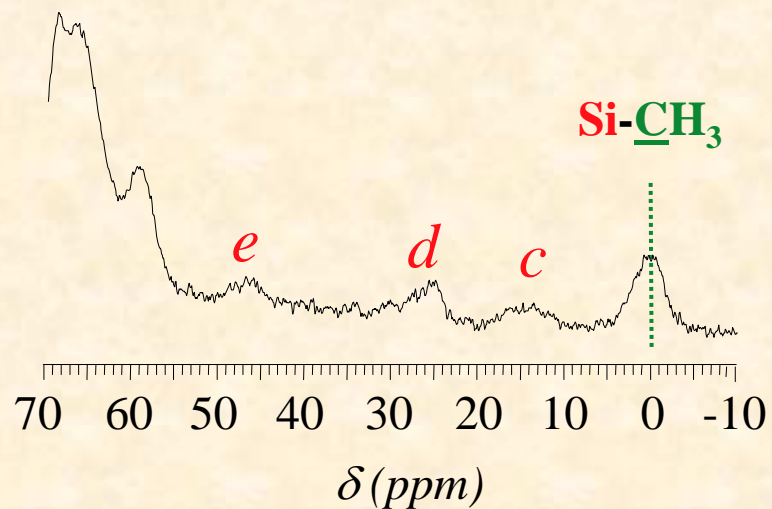
□ Spectroscopic characterizations



FTIR

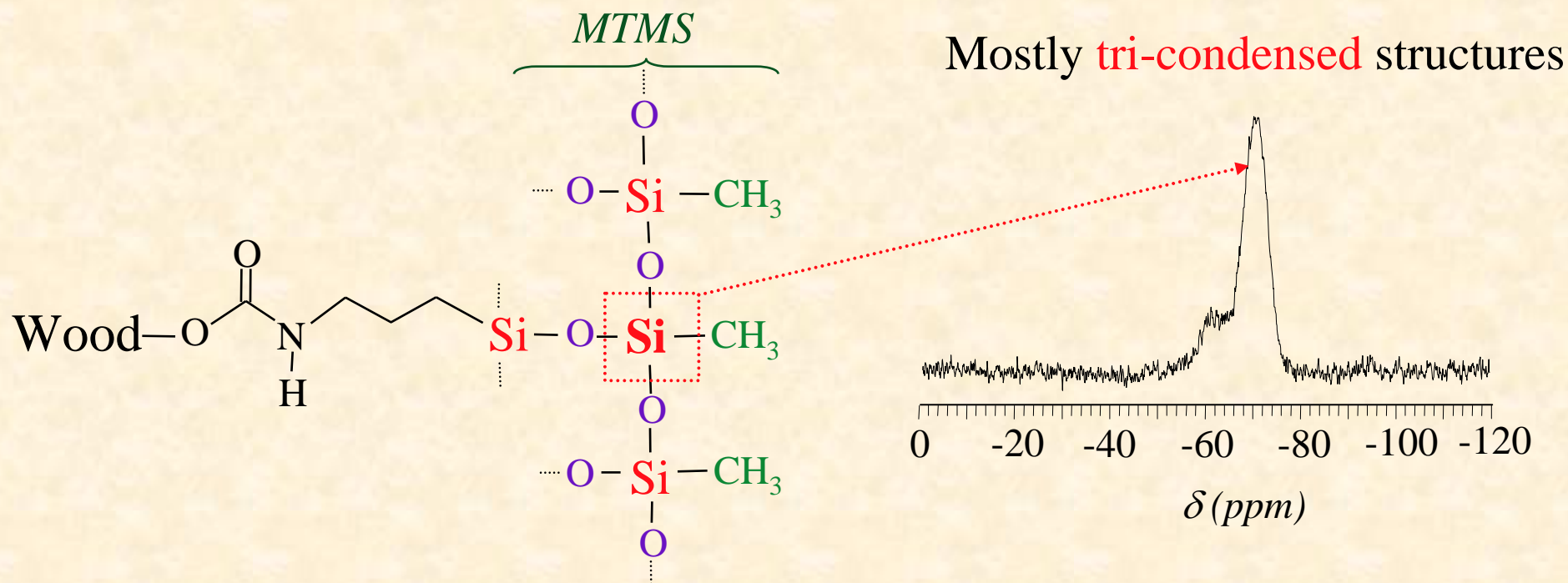


¹³C solid NMR



COUPLING WITH MTMS

□ ^{29}Si CP-MAS NMR spectroscopy



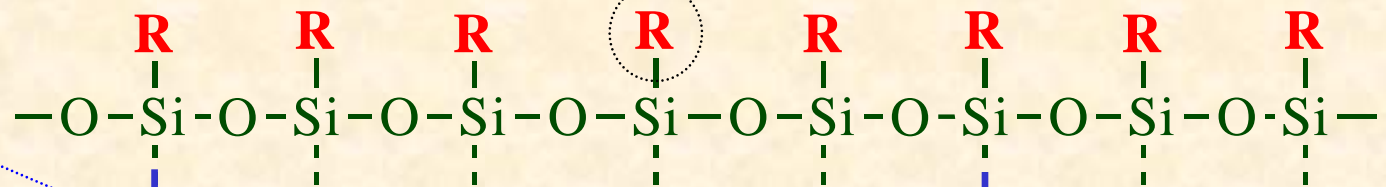
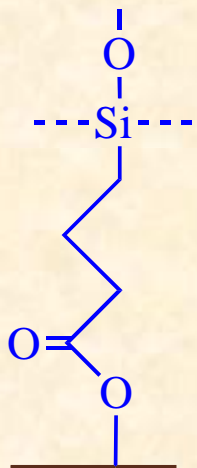
- **MTMS** grafted in wood in the form of a tri-condensed polysiloxane
- Grafting highly resistant to water leaching

PERSPECTIVES

□ Application to WPC's

- Hydrophilic
- Hydrophobic
- Organoreactive
- UV or heat resistant...

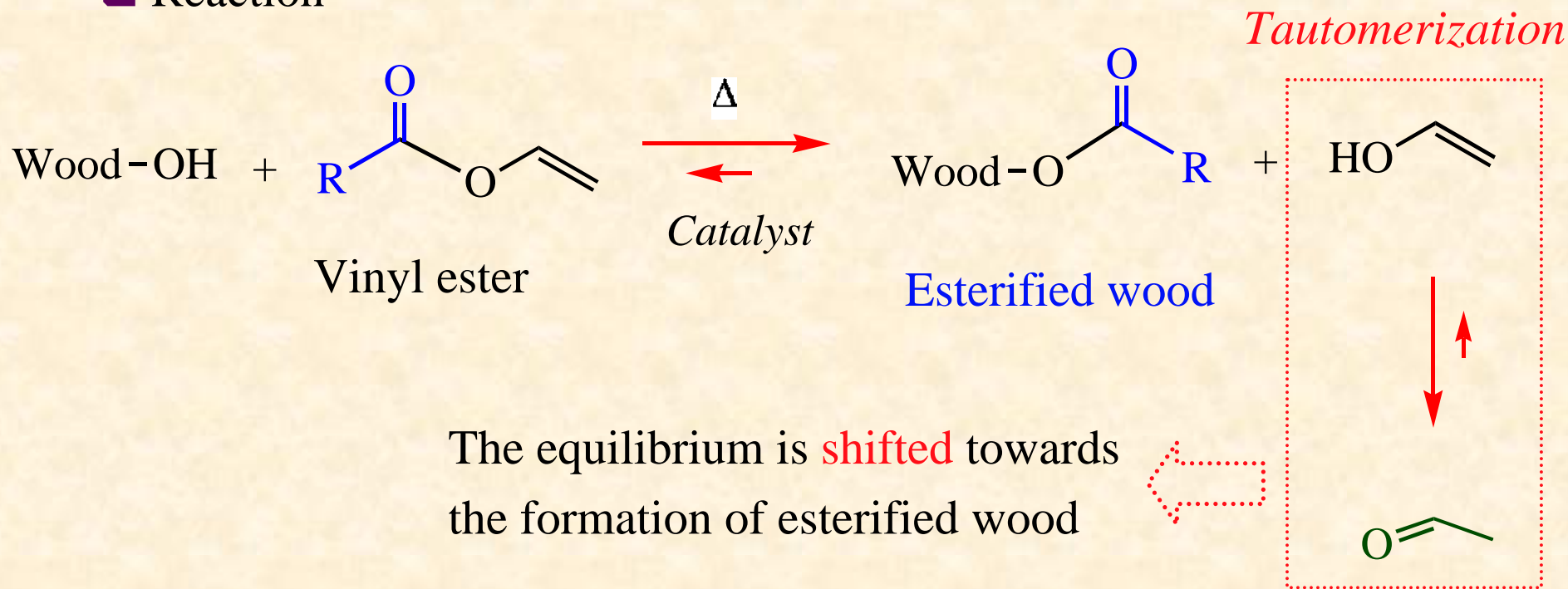
Anchoring site
(coupling agent)



Lignocellulosic fiber

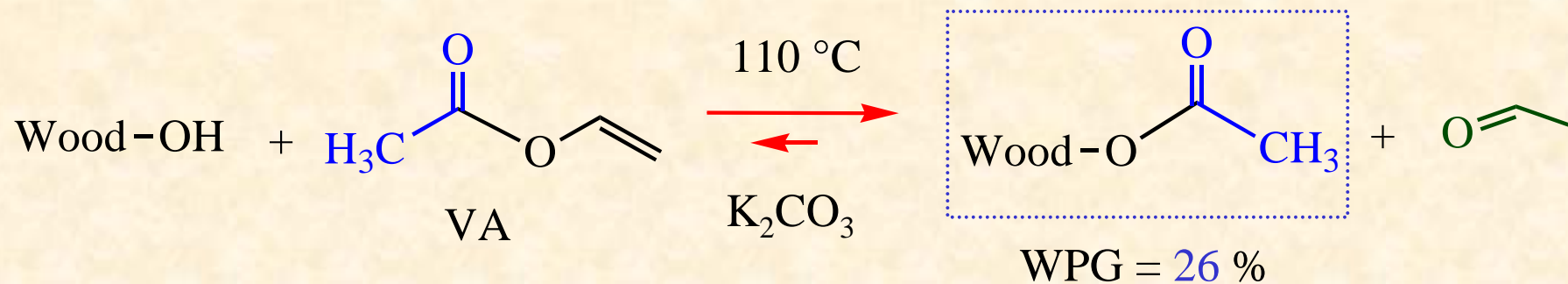
TRANSESTERIFICATION WITH VINYL ESTERS

□ Reaction



- The **acetaldehyde** formed as a by-product can be easily removed after reaction ($\text{bp}_{760} = 21^\circ\text{C}$)

REACTION WITH VINYL ACETATE

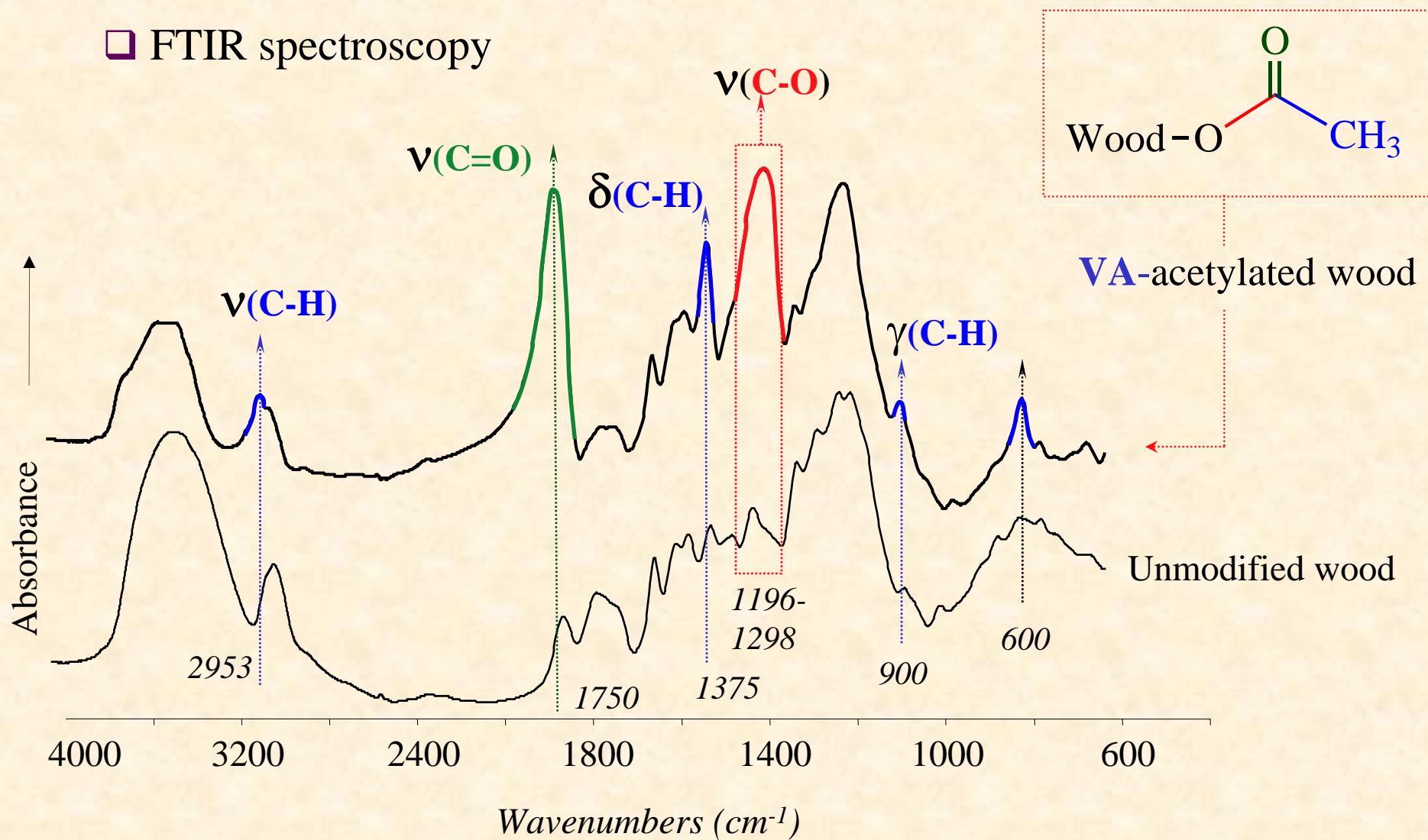


➤ Preliminary experiments were performed with **sawdust**, under a standard set of conditions:

- 1.1 mmol K_2CO_3 /g wood, as catalyst
- Dimethylformamide (DMF) as solvent
- Reaction for 3 hours, at 110 °C

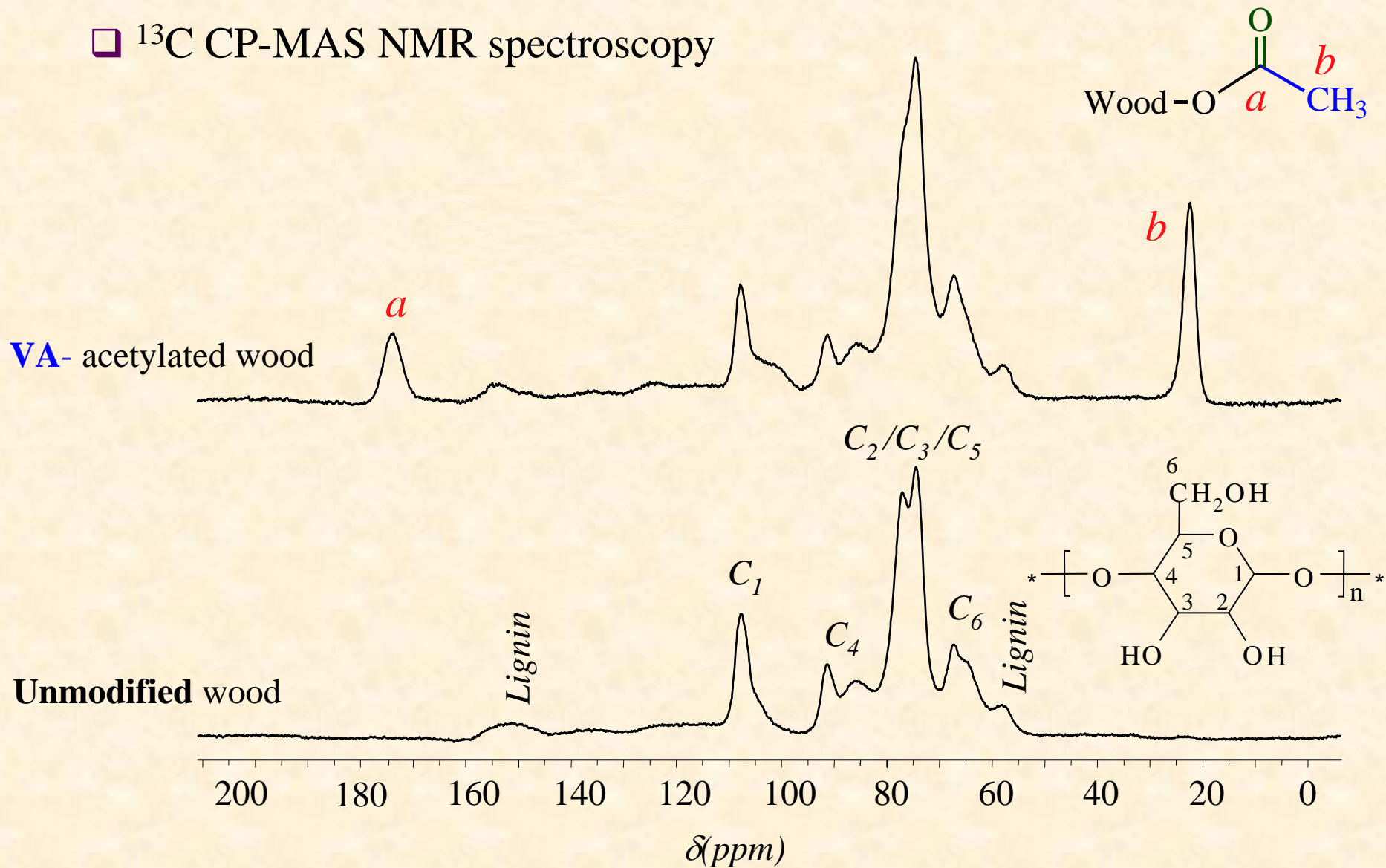
REACTION WITH VINYL ACETATE

□ FTIR spectroscopy



REACTION WITH VINYL ACETATE

□ ^{13}C CP-MAS NMR spectroscopy



REACTION WITH VINYL ACETATE

Temperature (°C)	25	50	90	110	130	150
WPG (%)	0.8	9.5	24.6	26.2	27.2	27.9

Reaction time = 3 h ; K_2CO_3 = 1.1 mmol/g wood

Reaction time (min)	10	30	60	180	360	960
WPG (%)	2.8	7.5	22.3	24.6	26.7	25.6

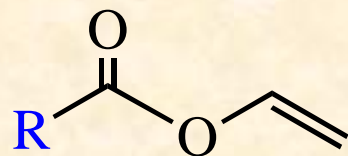
Temperature = 90 °C ; K_2CO_3 = 1.1 mmol/g wood

K_2CO_3 (mmol/g wood)	0	0.05	0.1	1.1	10.7	21
WPG (%)	0.0	0.05	5.7	24.6	24.8	26.9

Reaction time = 3 h ; Temperature = 90 °C

REACTION WITH OTHER VINYL ESTERS

□ Saturated vinyl esters



-R

WPG (%)

Vinyl propionate

-CH₂-CH₃

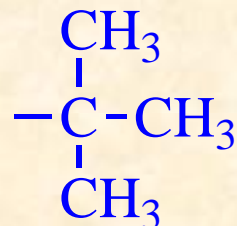
23.6

Vinyl butyrate

-CH₂-CH₂-CH₃

30.0

Vinyl pivalate



15.2

Vinyl dodecanoate

-CH₂-(CH₂)₇-CH₃

49.6

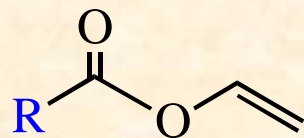
Vinyl stearate

-CH₂-(CH₂)₁₅-CH₃

52.2

REACTION WITH OTHER VINYL ESTERS

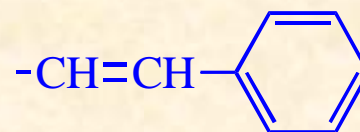
□ Unsaturated vinyl esters



-R

WPG (%)

Vinyl cinnamate



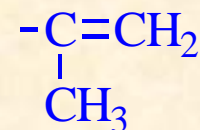
46.3

Vinyl crotonate



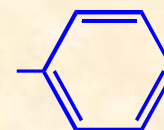
16.8

Vinyl methacrylate



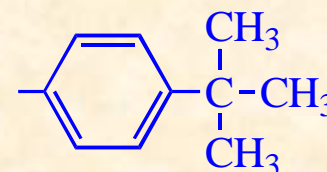
24.8

Vinyl benzoate

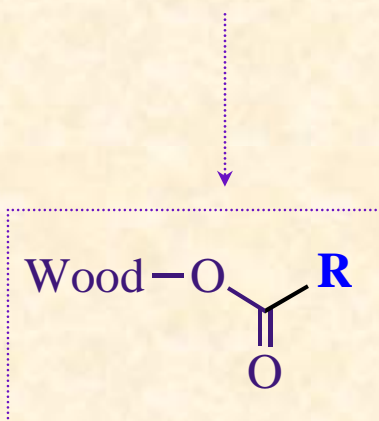


46.5

Vinyl 4-*tert*-butylbenzoate

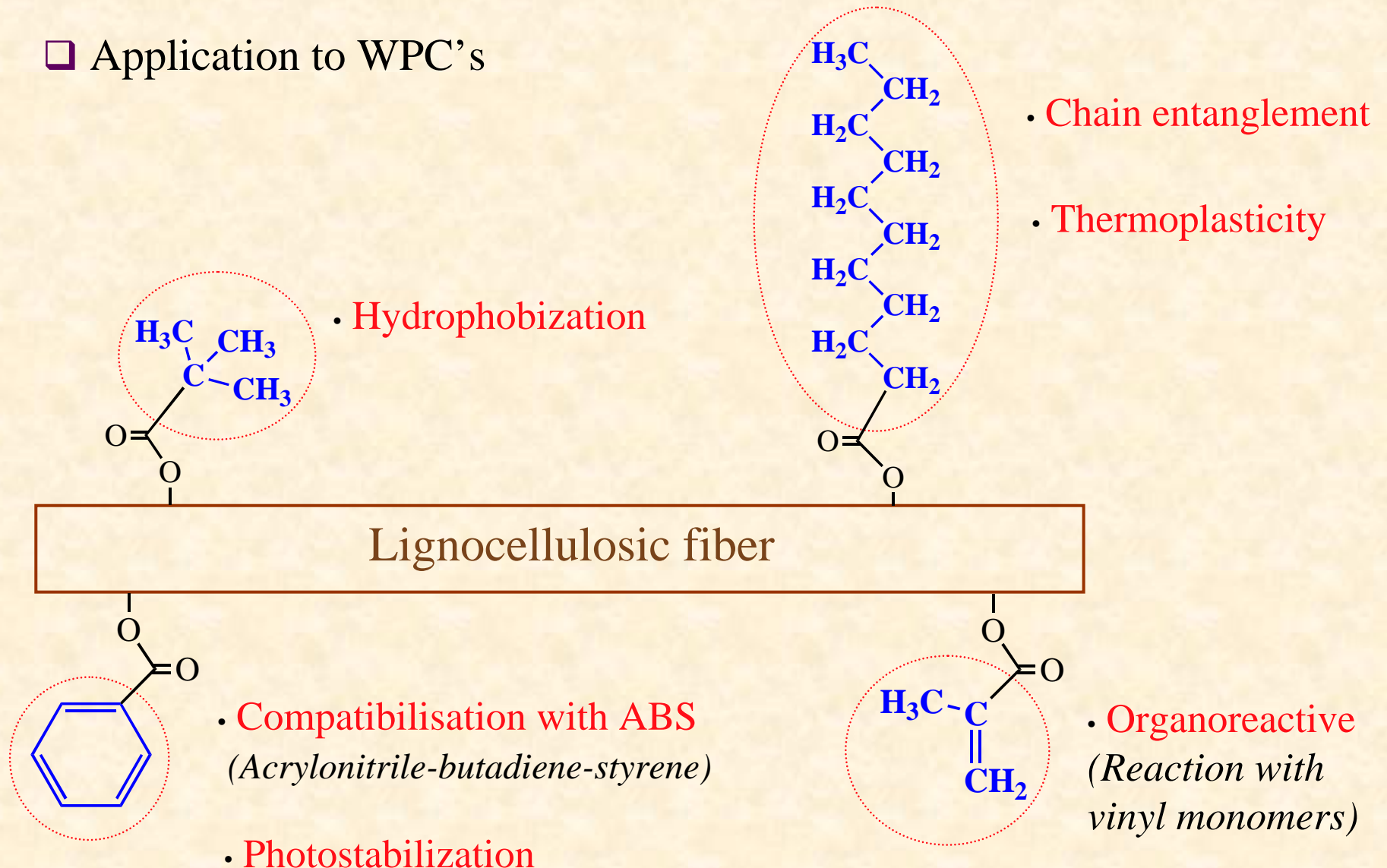


31.8



PERSPECTIVES

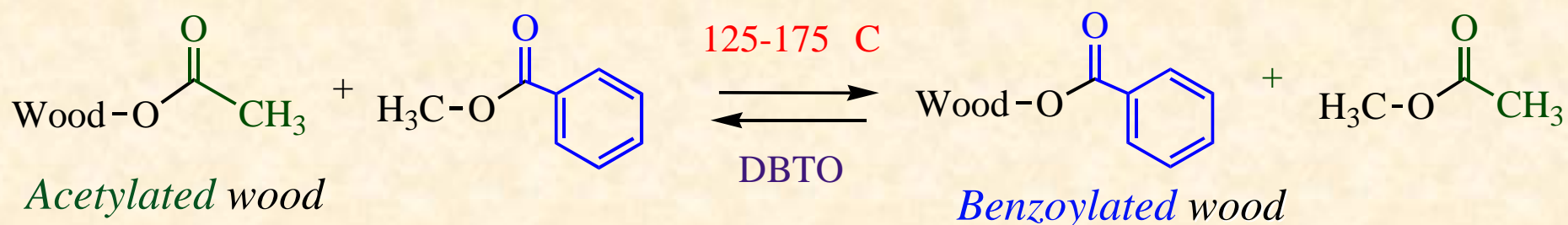
□ Application to WPC's



OTHER *TRANSESTERIFICATION* REACTIONS

□ Ester/ester exchange

DBTO = Dibutyltin oxide



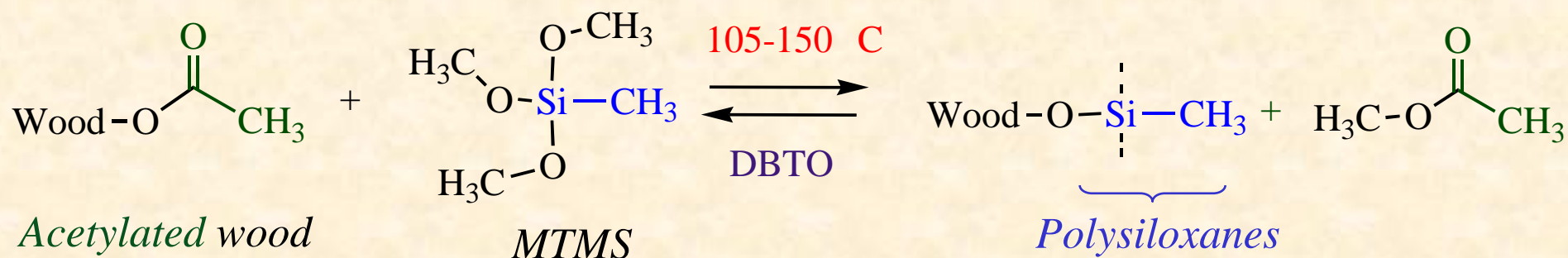
- The interchange reaction has been confirmed by:
 - Infrared spectroscopy (FTIR)
 - ¹³C solid-state NMR spectroscopy

- No reaction with un-acetylated wood

OTHER *TRANSESTERIFICATION* REACTIONS

DBTO = Dibutyltin oxide

□ Ester/alkoxysilane exchange



➤ The interchange reaction has been confirmed by:

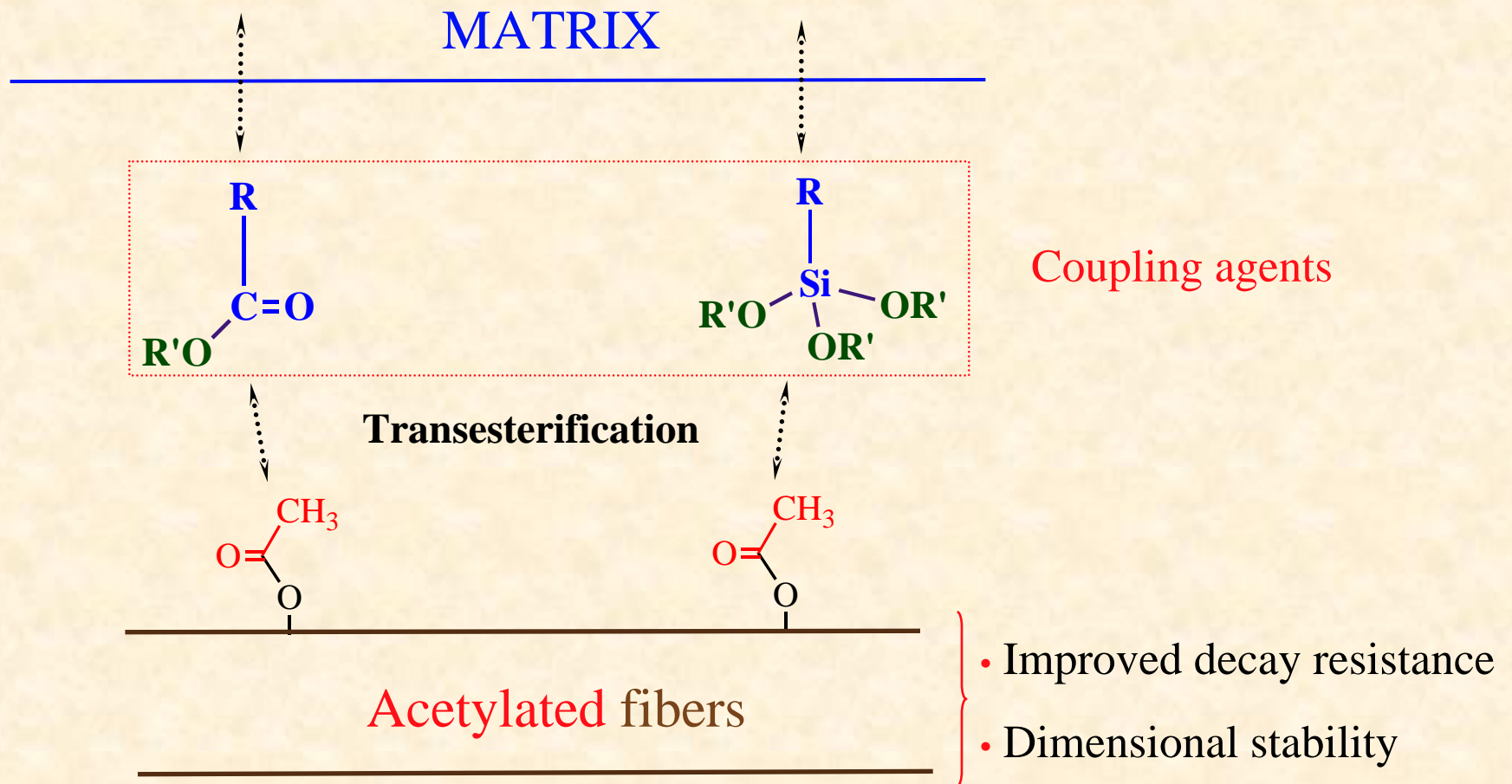
- Infrared spectroscopy (**FTIR**)

- **¹³C** and **²⁹Si** solid-state **NMR** spectroscopy

➤ No reaction with un-acetylated wood

PERSPECTIVES

- Application to WPC's from **acetylated** fibers



CONCLUSIONS

- ❑ Chemical **functionalization** of lignocellulosic fibers is a promising route to improve interfacial adhesion in WPC's
- ❑ Three novel **chemical routes** have been proposed to **durably graft** different functionalities in the fibers or at their surface:

All lignocellulosic materials

- Reaction with **alkoxysilanes** via an anchoring agent
- **Transesterification** with vinyl esters

Acetylated fibers

- **Ester/ester** or **alkoxysilane/ester** interchange reactions

CONCLUSIONS

- ❑ The **feasibility** of these methods has been confirmed by:
 - Weight Percent Gain calculation (**WPG**)
 - Infrared spectroscopy (**FTIR**)
 - Solid-state NMR spectroscopy (**^{13}C** and **^{29}Si** CP-MAS)

- ❑ These modifications will now be used to prepare **model interfaces**, at a laboratory scale, and **modelize** the mechanisms involved in the interfacial region of WPC's

ACKNOWLEDGMENTS

□ Ph.D. students and post-docs

- **Dr Philippe Tingaut:** Post-Doctoral Associate at the University of Tennessee
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- **Mohamed Jebrane:** Ph.D. Student at the University of Bordeaux

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Thank you for your attention!